# Cobalt-Catalyzed Cross-Benzannulation of Conjugated Enynes and Diynes

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**S** Supporting Information

[AB](#page-5-0)STRACT: The  $[4 + 2]$  cross-benzannulation of conjugated enynes and diynes under cobalt-catalysis led to 1,2,3-trisubstituted benzene derivatives in good yields. The reaction proceeds smoothly in absolute regiospecific control when symmetrical diynes are applied. Moreover, the use of unsymmetrical diynes was investigated, resulting in the formation of the unprecedented regioisomers as major products, which is in contrast to the results obtained in palladium-catalyzed benzannulation reactions. Also, 4-



bromophenyl-substituted starting materials could be applied successfully in the cobalt-catalyzed process, which can be problematic in the palladium-catalyzed counterpart.

The regioselective construction of polysubstituted benzene<br>derivatives represents an outstanding topic among organic<br>reactions, and innumerable surtheses have been developed in reactions, and innumerable syntheses have been developed in the recent years.<sup>1</sup> Among those reactions, one of the most prominent approaches can be considered to be the  $[2 + 2 + 2]$ cyclotrimerizatio[n](#page-5-0) of alkynes catalyzed by several transition metals.<sup>2</sup> Further approaches are the Diels−Alder reaction of conjugated 1,3-dienes with alkynes,<sup>3</sup> the dehydro-Diels-Alder reactio[n](#page-5-0),<sup>4</sup> and the  $\lceil 4 + 2 \rceil$  benzannulation of conjugated enynes and diynes (Scheme 1). In contrast [to](#page-5-0) the Diels−Alder reaction, benzann[u](#page-5-0)lation leads directly to benzene derivatives, requiring no further oxidation of the 1,4-dihydroaromatic compounds.

## Scheme 1. Palladium- and Cobalt-Catalyzed Cross-Benzannulation



Two transition metal-catalyzed benzannulation reactions of enynes and diynes have been reported thus far: first, the palladium-catalyzed homobenzannulation<sup>5</sup> of two conjugated enynes 1 to symmetric styrenes 2 as well as the crossbenzannulation of enynes 1 and diynes 4 r[ep](#page-5-0)orted by Yamamoto and Gevorgyan;<sup>6</sup> second, the regiodivergent cobalt-catalyzed homobenzannulation of enynes 1 recently reported by our group.<sup>7</sup> In the [co](#page-5-0)balt-catalyzed reaction, the regioselectivity is strongly solvent-dependent leading to regioisomer 3, which is not accessible via the palladium-catalyzed reaction.

On the basis of the results in the regiodivergent cobaltcatalyzed homobenzannulation, we extended our study toward the cross-benzannulation of conjugated enynes 1 with symmetrical as well as unsymmetrical diynes of type 4. To probe the feasibility of the reaction, we initially investigated the reaction of but-3-en-1-yn-1-ylbenzene 7a with the symmetric 1,4-diphenylbuta-1,3-diyne 8a in the presence of  $\text{CoBr}_2(\text{dppp})$  (10 mol %),<sup>8</sup> zinc powder (20 mol %), and zinc iodide (20 mol %) in dichloromethane at ambient temperatures (Scheme 2). Th[e](#page-5-0)





reaction led to the symmetric 1,2,3-trisubstituted benzene derivative 9a as a single isomer in an excellent yield of 90%. Besides the desired product, the formation of the homobenzannulation reaction of enyne 7a and cyclotrimerization of diyne 8a was also observed (∼10%).

In contrast to the homobenzannulation, no solvent dependency in terms of regiocontrol could be observed. The use of CH<sub>3</sub>CN, THF, DMF, or DMSO led to inferior conversions or unselective homobenzannulation, and cyclotrimerization of the starting materials was encountered. Moreover, reaction temperatures below 15 °C slowed the reaction dramatically, whereas temperatures above 35 °C led to cyclotrimerization preferentially. The use of 1- and 2-substituted enynes was tested, but

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neither homobenzannulation nor cross-benzannulation occurred with these types of substrates.

With the optimized reaction conditions in hand, we first focused our efforts toward the cross-benzannulation of 4 substituted enynes 7 and symmetric diynes 8 (Scheme 3).

## Scheme 3. Cobalt-Catalyzed Cross-Benzannulation Using Conjugated Enynes and Symmetric Diynes



The present reaction tolerated a wide range of enynes with electron-neutral as well as electron-rich and -deficient arenes, affording corresponding terphenyls 9b−9e in yields between 42 and 80% (Table 1, entries 1−4). Alkyl-substituted enynes gave products 9f and 9g in moderate yields after a prolonged reaction time (Table 1, entries 5 and 6) and applying either  $CoBr<sub>2</sub>(dppp)$  or  $CoBr<sub>2</sub>(dppe)$  as precatalyst.<sup>8</sup> Also, the sterically hindered thienyl-substituted enyne (Table 1, entry 7) gave product 9h in a good yield of 56%, and [i](#page-5-0)ts structure could be verified by X-ray analysis (see Supporting Information).

Moreover, various diynes 8i−8k with [electron-donating as](#page-5-0) [well](#page-5-0) as electron-withdrawing substituents could be applied in good yields (Table 1, entries 8−10). In the case of 9j, less cyclotrimerization than that with 9i took place, presumably due to a more sterically hindered diyne moiety. Also of particular interest is bromo-substituted terphenyl 9l, which could be synthesized in good yield (Table 1, entry 11). In contrast to the palladium-catalyzed benzannulation reactions, under the cobaltcatalyzed reaction conditions, the bromo substituent remained unchanged; neither proto-debromination nor other crosscoupling products could be observed. Accordingly, the scope of the reaction could be expanded so that further modifications will be possible. To our regret, terminal diynes, of the  $R$ -C $\equiv$ C- $C\equiv$  CH type, as well as diynes with electron-withdrawing substituents led predominantly to the corresponding cyclotrimerization products and led to the desired benzannulation products only in trace amounts.

Moving onward in the investigation, we focused our attention on the regioselective benzannulation reaction utilizing conjugated enynes and unsymmetrical diynes (Scheme 4).

In a first survey applying but-3-en-1-yn-1-ylbenzene 7a with nona-1,3-diyn-1-ylbenzene 4a, products 10a and 1[1a](#page-2-0) could be obtained in 50% yield, and a moderate regioselectivity of 10a/ 11a (61:39) was detected (Table 2, entry 1). Thereby, terphenyl 10a containing the phenyl substituent at the annulated ring was predominantly formed. An incr[ea](#page-2-0)sed regioselectivity could be observed when TMS-substituted diynes 4b−4f were used. Products 10b, 10c, and 10f could be obtained in 46−60% yield and good regioselectivity (Table 2, entries 2, 3, and 7). Noteworthy is that in the case of the ortho- and paramethoxyphenyl-substituted diynes 4[d](#page-2-0) and 4e, terphenyls 10d and 10e could be isolated in 61−69% yield and excellent regioselectivity (Table 2, entries 4 and 5). The divergent nature of the cobalt- versus palladium-catalyst system becomes obvious when diyne 4e is app[lie](#page-2-0)d. The palladium-catalyzed benzannulation of 4e resulted in the exclusive formation of regioisomer

no.	product $(9)^c$	t/h	yield $^a$
$\mathbf{1}$	1 I 9b	$1\,8$	42%
$\overline{\mathbf{c}}$	MeO 9c	$22\,$	$64\%$
3	$E$ tO <sub>2</sub> C 9d	$1\,8$	$80\%$
$\overline{\mathcal{L}}$	NC 9e	48	43%
5	<b>BnO</b> 9f	48	$45\%$ $^b$
6	Me <sup>2</sup> 10 9g	48	45% b
$\overline{7}$	Me 9h	15	56%
8	II OMe 91 ÓМе	$\frac{14}{5}$	62%
9	OMe 9j	14	72%
10	CO <sub>2</sub> Et $_{\rm CO_2Et}^{\rm I}$ 9k	16	66%
$\overline{11}$	Br OMe ∫ OMe 91	<b>20</b>	60%

a Reaction conditions: enyne 7 (1.2 equiv), diyne 8 (1.0 equiv),  $CoBr_2(dppp)$  (10 mol %), Zn powder (20 mol %), ZnI<sub>2</sub> (20 mol %), COLL2, rt.  $\text{CoBr}_2(\text{dpp})$  or  $\text{CoBr}_2(\text{dpp})$  or  $\text{CoBr}_2(\text{dpp})$  can be used as precatalyst.<br>CH<sub>2</sub>Cl<sub>2</sub>, rt.  $\text{CoBr}_2(\text{dpp})$  or  $\text{CoBr}_2(\text{dpp})$  can be used as precatalyst. <sup>c</sup>In all reactions, trimerization of the enyne or diyne could be observed (∼5−10%).

<span id="page-2-0"></span>Scheme 4. Cobalt-Catalyzed Cross-Benzannulation Applying Unsymmetrical Diynes



Table 2. Cobalt-Catalyzed Benzannulation of Conjugated Enynes with Unsymmetrical Diynes<sup>a</sup>



a Reaction conditions: enyne 7a (1.8 equiv), diyne 4 (0.5 mmol, 1.0 equiv),  $Co(dppp)Br_2$  (10 mol %), Zn powder (20 mol %), ZnI<sub>2</sub> (20 mol %), CH<sub>2</sub>Cl<sub>2</sub>, rt, 48 h.  $^{b}$ Enyne 7a (1.2 equiv), diyne 4 (1.0 equiv),  $Pd(PPh<sub>3</sub>)$ 4 (5 mol %), NEt<sub>3</sub> (2.0 equiv), toluene, 100 °C, 48 h. <sup>c</sup>The regioisomer ratio was determined by GC analysis.  ${}^{d}$ In all reactions, trimerization of the enyne or diyne could be observed (∼5−10%).

11e (Table 2, entry 6) with the TMS group at the annulated ring as a single isomer in good yield (Table 2, entry 6). These results illustrate the different behavior of the cobalt- and palladiumcatalyst systems and the complementary of the catalyst systems

Finally, we would like to direct attention to product 10f, which is difficult to generate by palladium-catalyzed benzannulation; the bromo-substituent will afford us the ability to direct further modifications, such as Ullman-type dimerization or cross-coupling reactions.

In conclusion, we described the first cobalt-catalyzed enynediyne cross-benzannulation reaction leading to 1,2,3-substituted benzene derivatives. The products could be obtained in absolute regiospecific control and good yields. Several functionalyzed enynes and diynes could be applied, illustrating the high functional group tolerance of this reaction. Furthermore, regioselective cross-benzannulation utilizing unsymmetrical diynes was investigated. The products were formed in moderate to good yields and good regioselectivities, giving the opposite

regioisomer from that with the palladium-catalyzed benzannulation reaction as the major product. Consequently, the cobaltcatalyzed transformation broadened the scope of this reaction and led to new products not easily accessible by other methods.

## **EXPERIMENTAL SECTION**

General Methods. All reactions requiring water- or air-sensitive compounds were carried out in vacuum- and flame-dried flasks utilizing Schlenk techniques under an argon atmosphere. DCM was died over P4O10, and THF was dried over sodium/benzophenone and distilled under a nitrogen atmosphere.  $ZnI_2$  was dried in vacuo at 150 °C prior to use. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded at 300 and 75 MHz, respectively, at room temperature utilizing preset pulse programs. The chemical shifts are given relative to tetramethylsilane as an internal standard. The solvent signal was used for calibration (<sup>1</sup>H NMR  $\delta$  $(CHCl<sub>3</sub>)$  = 7.26 ppm; <sup>13</sup>C NMR  $\delta$  (CHCl<sub>3</sub>) = 77.16 ppm). The following abbreviations are used to describe the signals: s (singlet), d (doublet), t (triplet), q (quartet), quin (quintet), sext (sextet), and m (multiplet). The coupling constants (J) are given in hertz (Hz). Infrared spectra (IR) were recorded on a FT-IR spectrometer. The absorption bands are given in wave numbers (cm<sup>−</sup><sup>1</sup> ). The detected ion masses  $(m/z)$  are reported in u corresponding to the intensity of the signals as a percentage of the most intense signal. GC analysis was performed on a gas chromatograph with a flame ionization detector (FID). High-resolution mass spectra (HRMS) were recorded as electron ionization spectra (EI/HRMS) utilizing a quadrupole mass analyzer at an energy of 70 eV or as electron spray ionization (ESI/ HRMS) using an LTQ-FT system. The detected ion masses  $(m/z)$  are given in u.

General Procedure for the Synthesis of Terminal 1,3-Enynes from Terminal Alkynes.  $PdCl_2(PPh_3)_2$  (2 mol %), CuI (2 mol %), and  $Ph_3P$  (4 mol %) were suspended in THF (2 mL/mmol). Degassed  $iPr_2NH$  or Et<sub>3</sub>N (2.30 equiv), the alkyne (1.00 equiv), and vinyl bromide solution (1 M in THF, 1.20 equiv) were added successively. The reaction mixture was stirred at room temperature until complete conversion was determined by TLC and GC-MS analyses. The reaction mixture was concentrated under reduced pressure, and  $H_2O$  was added followed by extraction with  $CH_2Cl_2$ . The organic layer was dried over MgSO4, filtered, and concentrated under reduced pressure. The crude product was purified by flash chromatography (silica, eluent: npentane/ $Et_2O$ ) to give the desired product.

Literature known compounds prepared by this method: but-3-en-1 yn-1-ylbenzene 7a. <sup>9</sup> Yield: 90% (1.15 g, 9.00 mmol). 2-(But-3-en-1-yn-1-yl)naphthalene 7b. <sup>7</sup> Yield: 66% (232 mg, 1.30 mmol). ((Pent-4-en-2- yn-1-yloxy)methyl[\)b](#page-5-0)enzene 7f.<sup>10</sup> Yield: 43% (297 mg, 7.72 mmol). Hexadec-1-en-3-yne [7](#page-5-0)g. <sup>11</sup> Yield: 82% (3.24 g, 14.7 mmol). 1-Bromo-4- (but-3-en-1-yn-1-yl)benzene 7h. <sup>12</sup> Yield: 78% (397 mg, 1.92 mmol).

Ethyl 4-(But-3-en-1-[yn](#page-5-0)-1-yl)[be](#page-5-0)nzoate 7d. Yield: 86% (579 mg, 2.89 mmol), orange oil. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.96–7.86 (m, 2H), 7.49−7.34 (m, 2H), 5.96 (dd, J = 17.6, 11.1 Hz, 1H), 5.70 (dt, J = 17.7, 2.1 Hz, 1H), 5.52 (dt,  $J = 11.1$ , 2.1 Hz, 1H), 4.30 (q,  $J = 7.1$  Hz, 2H), 1.32 (t, J = 7.1 Hz, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  166.2, 131.6, 130.1, 129.6, 128.0, 127.9, 117.0, 91.0, 89.4, 61.3, 14.5. IR (ATR) 2982, 1714, 1607, 1404, 1367, 1265, 1173, 1100, 1019, 969, 922, 854, 766, 693. HR MS (EI+)  $m/z$  calcd for  $C_{13}H_{12}O_2$  ([M]<sup>+</sup>), 200.0837; found, 200.0824.

4-(But-3-en-1-yn-1-yl)benzonitrile 7e. Yield: 86% (579 mg, 2.89 mmol), white solid. mp 63–64 °C. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$ 7.60−7.57 (m, 2H), 7.51−7.46 (m, 2H), 6.02 (dd, J = 17.5, 11.1 Hz, 1H), 5.79 (dd, J = 17.6, 2.1 Hz, 1H), 5.63 (dd, J = 11.1, 2.1 Hz, 1H).  ${}^{13}C{^1H}$  NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  132.1, 132.1, 128.8, 128.2, 118.5, 116.6, 111.7, 92.4, 88.3. IR (ATR) 3883, 2223, 2153, 2073, 2020, 1979, 1921, 1774, 1717, 1602, 1499, 1442, 1402, 1271, 1179, 1105, 1014, 956, 924, 835, 773, 731, 694, 618, 554, 472, 430. MS (EI+)  $m/z = 153$  (100, [M] <sup>+</sup> ), 147 (20), 139 (36), 130 (48), 127 (60), 102 (20), 75 (24), 69 (24). HR MS (EI+)  $m/z$  calcd for  $C_{11}H_7N$  ([M]<sup>+</sup>), 153.0578; found, 153.0570.

General Procedure for the Synthesis of Symmetrically Substituted Buta-1,3-diynes from Acetylenes. Under an oxygen atmosphere, copper(II) chloride dihydrate (0.10 equiv) and N,N,N′,N′-tetramethylethylenediamine (0.30 equiv) were dissolved in  $CH_2Cl_2$  (1 mL/mmol). The alkyne (1.00 equiv) was added, and the mixture was stirred at room temperature until complete conversion was determined by TLC and GC-MS analyses.  $H_2O$  was added, and the organic phase was separated, followed by extraction with  $CH_2Cl_2$ . The combined organic phases were dried over MgSO<sub>4</sub> and filtered, and the solvent was removed under reduced pressure. The crude product was purified by flash chromatography (silica, eluent:  $n$ -pentane/Et<sub>2</sub>O).

Literature known compounds prepared by this method: 1,4 diphenylbuta-1,3-diyne 8a. <sup>13</sup> Yield: 98% (993 mg, 4.91 mmol). 1,4- Bis(4-methoxyphenyl)buta-1,3-diyne 8b. <sup>14</sup> Yield: 79% (1.04 g, 3.95 mmol). 1,4-Bis(2-methoxy[ph](#page-5-0)enyl)buta-1,3-diyne 8c. <sup>15</sup> Yield: 55% (721 mg, 2.75 mmol). Diethyl 4,4′-(buta-1,3-d[iyn](#page-5-0)e-1,4-diyl)dibenzoate 8d. 6l Yield: 58% (311 mg, 0.90 mmol).

General Procedure for the Synthesis of [Un](#page-5-0)symmetricall[y](#page-5-0) **Substituted Diynes.** Following a modified procedure of Tykwinsky, alkyne (1.25 equiv) was dissolved in  $Et<sub>2</sub>O$  (2 mL/mmol) and cooled to −78 °C. Then, n-BuLi (2.5 M in hexane, 1.20 equiv) was added, and t[he](#page-5-0) solution was allowed to warm to room temperature over a period of 30 min. The appropriate aldehyde (1.00 equiv) was added, and the solution was stirred at room temperature until complete conversion was determined by TLC and GC-MS analyses. The reaction was quenched with saturated NH4Cl solution, the organic phase was separated and washed with  $3 \times 15$  mL brine, and the aqueous phase was extracted with  $3 \times 15$  mL Et<sub>2</sub>O. The combined organic phases were dried over MgSO4, and the solvent was removed in vacuo. The thus formed alcohol (1.00 equiv) was dissolved in  $CH_2Cl_2$  (4 mL/mmol), PCC (1.10 equiv) was added, and the solution was stirred at room temperature until completion was determined by TLC and GC-MS analyses. The solution was filtered over a plug of silica, and the solvent was removed. PPh<sub>3</sub> (3.60 equiv) was dissolved in  $CH_2Cl_2$  (3 mL/ mmol),  $CBr_4$  (1.80 equiv) was added slowly at 0 °C, and the solution was stirred for 15 min. Then, the ketone (1.00 equiv) was added, and the mixture was stirred for 2 h at room temperature. The mixture was concentrated, n-pentane was added, the mixture was filtered over a plug of silica (*n*-pentane/CH<sub>2</sub>Cl<sub>2</sub> 4:1), and the solvent was removed. The dibromo-olefin was dissolved in benzene (1 mL/mmol), and hexane was added (3 mL/mmol). The mixture was cooled to  $-35$  °C, and n-BuLi (2.5 M in hexane, 1.00 equiv) was added. The mixture was allowed to warm to room temperature and was stirred until completion was determined by GC-MS analysis. The reaction was quenched with a saturated NH<sub>4</sub>Cl solution, the organic phase was separated and washed with  $3 \times 15$  mL brine, and the aqueous phase was extracted with  $3 \times 15$ mL Et<sub>2</sub>O. The combined organic phases were dried over MgSO<sub>4</sub>, and the solvent was removed under reduced pressure. The crude product was purified by flash chromatography (silica, eluent:  $n$ -pentane/Et<sub>2</sub>O).

Literature known compounds prepared by this method: nona-1,3 diyn-1-ylbenzene 4a. <sup>17</sup> Yield: 32% (576 mg, 3.16 mmol). Trimethyl- (phenylbuta-1,3-diyn-1-yl)silane  $4b.^{16}$  Yield: 70% (1.38 g, 6.96 mmol). Trimethyl(nona-1,3-[diy](#page-5-0)n-1-yl)silane 4c. <sup>16</sup> Yield: 55% (1.06 g, 5.51 mmol). ((2-Methoxyphenyl)buta-[1,3](#page-5-0)-diyn-1-yl)trimethylsilane 4d.<sup>16</sup> Yield: 53% (1.20 g, 5.25 mmol). ((4-M[eth](#page-5-0)oxyphenyl)buta-1,3-diyn-1 yl)trimethylsilane 4e. <sup>16</sup> Yield: 75% (1.20 g, 7.47 mmol). (([4-](#page-5-0) Bromophenyl)buta-1,3-diyn-1-yl)trimethylsilane 4f. <sup>18</sup> Yield: 70% (1.88 g, 6.79 mmol).

General Procedu[re](#page-5-0) for the Cobalt-Catalyzed [Cr](#page-5-0)oss-Benzannulation of 1,3-Enynes with Symmetrical Buta-1,3-diynes. Under argon atmosphere cobaltdibromo(1,3-bis(diphenylphosphino) propane)  $[= \text{CoBr}_2(\text{dppp})]$  (10 mol %), zinc powder (10 mol %), and zinc iodide (10 mol %) were dissolved in  $CH_2Cl_2$  (2 mL/mmol). The 1,3-enyne (0.60 mmol, 1.20 equiv) and buta-1,3-diyne (0.50 mmol, 1.00 equiv) were added subsequently. The reaction was stirred at room temperature until complete conversion was determined by TLC and GC-MS analyses. After complete reaction, n-pentane was added followed by filtration over a plug of silica (eluent:  $n$ -pentane/Et<sub>2</sub>O 1:1). The solvent was removed, and the crude mixture was purified by flash chromatography (silica, eluent: *n*-pentane/ $Et_2O$ ).

2'-(Phenylethynyl)-1,1':3',1"-terphenyl 9a.<sup>6e</sup> Yield: 90% (149 mg, 0.45 mmol), colorless oil. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.73–7.65  $(m, 4H)$ , 7.53–7.35  $(m, 9H)$ , 7.22–7.15  $(m, 3H)$ , 7.00  $(dd, J = 6.4, 3.2$ Hz, 2H). <sup>13</sup>C{<sup>1</sup>H} NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  144.9, 141.0, 130.9, 129.6, 128.3, 128.0, 128.0, 127.8, 127.7, 127.3, 123.5, 120.2, 95.8, 88.9.

2-(2-(Phenylethynyl)-[1,1′-biphenyl]-3-yl)naphthalene 9b. Yield: 42% (80 mg, 0.21 mmol), slight yellow solid. mp 58−60 °C. <sup>1</sup> H NMR  $(300 \text{ MHz}, \text{CDCl}_3)$   $\delta$  8.16 (s, 1H), 7.92 (dt, J = 12.1, 8.8 Hz, 3H), 7.71  $(d, J = 7.7 \text{ Hz}, 2H), 7.60 - 7.35 \text{ (m, 7H)}, 7.32 - 7.20 \text{ (m, 2H)}, 7.20 - 7.06 \text{)}$ (m, 3H), 6.94 (d, J = 7.1 Hz, 2H). <sup>13</sup>C{<sup>1</sup>H} NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$ 145.3, 144.9, 141.3, 138.7, 133.4, 132.9, 131.6, 131.2, 130.6, 129.9, 128.9, 128.7, 128.4, 128.3, 128.2, 128.1, 127.9, 127.8, 127.6, 127.2, 126.2, 126.2, 123.6, 120.6, 96.3, 89.1. IR (ATR) 2962, 1413, 1259, 1085, 1015, 864, 795, 699, 662. MS (EI+)  $m/z = 380$  (68, [M]<sup>+</sup>), 376 (17), 363 (12), 350 (6), 303 (100), 276 (3), 182 (7), 151 (10). HR MS (EI+)  $m/z$  calcd for  $C_{30}H_{20}$  ([M]<sup>+</sup>), 380.1565; found, 380.1565.

4-Methoxy-2′-(phenylethynyl)-1,1′:3′,1″-terphenyl 9c. Yield: 64% (115 mg, 0.32 mmol), yellow oil. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.75− 7.62 (m, 4H), 7.53−7.36 (m, 6H), 7.25−7.15 (m, 3H), 7.12−6.98 (m, 4H), 3.90 (s, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  159.3, 145.3, 144.6, 141.4, 133.7, 131.2, 131.0, 129.8 (2C), 128.5, 128.2, 128.2, 128.0, 127.9, 127.5, 123.8, 120.3, 113.4, 95.9, 89.3, 55.5. IR (ATR) 3055, 2956, 2931, 2834, 1608, 1512, 1491, 1455, 1441, 1290, 1245, 1176, 1029, 910, 835, 802, 755, 691. MS (EI+) m/z = 360 (98, [M]<sup>+</sup> ), 329 (81), 313 (62), 300 (22), 289 (24), 283 (64), 268 (37), 252 (32), 239 (100), 213 (8), 187 (6), 163 (17), 157 (28), 150 (20). HR MS (EI+) m/z calcd for  $C_{27}H_{20}O$  ([M]<sup>+</sup>), 360.1514; found, 360.1496.

Ethyl 2′-(Phenylethynyl)-[1,1′:3′,1″-terphenyl]-4-carboxylate 9d. Yield: 80% (162 mg, 0.40 mmol), white solid. mp 89−91 °C. <sup>1</sup>H NMR (300 MHz, CDCl3) δ 8.13−8.03 (m, 2H), 7.73−7.64 (m, 2H), 7.64− 7.56 (m, 2H), 7.43−7.29 (m, 6H), 7.15−7.09 (m, 3H), 6.97−6.87 (m, 2H), 4.35 (q, J = 7.1 Hz, 2H), 1.35 (t, J = 7.1 Hz, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR  $(75 \text{ MHz}, \text{CDCl}_3)$  δ 166.8, 145.8, 145.4, 143.9, 141.1, 131.2, 129.8, 129.8, 129.6, 129.2, 129.2, 128.5, 128.3, 128.3, 128.3, 128.0, 127.7, 123.5, 120.4, 96.5, 88.6, 61.1, 14.5. IR (ATR) 3055, 2989, 2924, 2856, 1702, 1605, 1449, 1270, 1175, 1270, 1175, 1100, 1018, 754, 694, 513. HR MS (EI+)  $m/z$  calcd for  $C_{29}H_{22}O_2$  ([M]<sup>+</sup>), 402,1620; found, 402.1629.

2′-(Phenylethynyl)-[1,1′:3′,1″-terphenyl]-4-carbonitrile 9e. Yield: 35% (46 mg, 0.13 mmol), yellow solid. mp 87−89 °C. <sup>1</sup> H NMR (300 MHz, CDCl<sub>3</sub>) δ 7.76−7.64 (m, 3H), 7.64−7.51 (m, 2H), 7.47−7.33 (m, 5H), 7.33−7.23 (m, 1H), 7.23−7.06 (m, 4H), 6.98−6.81 (m, 2H). (m, 5H), 7.33−7.23 (m, 1H), 7.23−7.06 (m, 4H), 6.98−6.81 (m, 2H). 13C{1 H} NMR (75 MHz, CDCl3) δ 146.0, 145.6, 143.0, 140.8, 131.8, 131.1, 130.6, 129.7, 129.6, 129.3, 128.5, 128.4, 128.2, 128.0, 127.8, 123.2, 120.4, 119.2, 111.4, 96.8, 88.2. IR (ATR) 3055, 2966, 2225, 1949, 1809, 1719, 1601, 1492, 1446, 1396, 1275, 1180, 1073, 1022, 970, 914, 843, 803, 753, 691, 604, 546, 519, 429. MS (EI+) m/z = 355 (100, [M]<sup>+</sup>), 352 (21), 277 (89), 73 (34), 57 (20). HR MS (EI+)  $m/z$  calcd for  $C_{27}H_{17}N$  ([M]<sup>+</sup>), 355.1361; found, 355.1352.

3-((Benzyloxy)methyl)-2-(phenylethynyl)-1,1′-biphenyl 9f. Yield: 43% (82 mg, 0.22 mmol), white solid. mp 25−27 °C. <sup>1</sup> H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.61–7.53 (m, 2H), 7.50 (d, J = 7.5 Hz, 1H), 7.41– 7.32 (m, 4H), 7.31−7.23 (m, 4H), 7.22−7.12 (m, 7H), 4.83 (s, 2H), 4.63 (s, 2H). <sup>13</sup>C{<sup>1</sup>H} NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  144.6, 141.0, 141.0, 138.6, 131.6, 131.4, 129.7, 128.7, 128.6, 128.4, 128.4, 128.3, 128.0, 127.8, 127.6, 126.5, 123.6, 120.4, 97.6, 86.9, 73.1, 71.0. IR (ATR) 3058, 3029, 2921, 2854, 1491, 1451, 1356, 1106, 1068, 753, 691. MS (EI+)  $m/z = 355$  (1, [M]<sup>+</sup>), 283 (37), 265 (100), 239 (27), 105 (8), 91 (15). HR MS (EI+)  $m/z$  calcd for  $C_{28}H_{22}O$  ([M]<sup>+</sup>), 374.1671; found, 374,1675.

3-Dodecyl-2-(phenylethynyl)-1,1′-biphenyl **9g**. Yield:  $45\%$  (82 mg, 0.22 mmol), yellow oil. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.68–7.58 (m, 2H), 7.48−7.35 (m, 3H), 7.31−7.19 (m, 8H), 2.93 (t, J = 7.8 Hz, 2H), 1.83–1.65 (m, 2H), 1.55–1.11 (m, 18H), 0.87 (t, *J* = 5.9 Hz, 3H).<br><sup>13</sup>C{<sup>1</sup>H} NMR (75 MHz, CDCl<sub>3</sub>) δ = 145.9, 144.7, 141.4, 131.3, 129.7, 128.9, 128.14, 128.08, 127.9, 127.8, 127.4, 127.1, 124.0, 121.2, 96.4, 88.1, 35.5, 32.1, 31.0, 29.9, 29.8, 29.85 (2C), 29.82 (2C), 29.5, 22.8, 14.3. IR (ATR): 3058, 2922, 2852, 1490, 1459, 1069, 1027, 911, 754, 698, 689. MS (EI+) m/z = 422 (14, [M]<sup>+</sup> ), 267 (60), 252 (51), 202 (35), 178 (38), 154 (9), 127 (14), 101 (16), 85 (29), 57 (100). HR MS (EI+)  $m/z$  calcd for  $C_{32}H_{38}$  ([M]<sup>+</sup>), 422.2974; found, 422.2967.

2-Methyl-5-phenyl-3-(2-(phenylethynyl)-[1,1′-biphenyl]-3-yl) thiophene 9h. Yield: 56% (119 mg, 0.28 mmol), yellow solid. mp 156−158 °C. <sup>1</sup> H NMR (300 MHz, CDCl3) δ 7.65−7.58 (m, 2H), 7.54−7.49 (m, 2H), 7.42−7.35 (m, 2H), 7.35−7.30 (m, 4H), 7.25 (ddd, J = 7.8, 4.8, 1.5 Hz, 3H), 7.17−7.12 (m, 1H), 7.09−7.00 (m, 3H), 6.98−6.91 (m, 2H), 2.37 (s, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$ 144.8, 141.0, 140.2, 139.2, 139.1, 136.1, 134.8, 133.5, 131.3, 129.8, 129.0, 128.6, 128.3, 128.1, 128.0, 127.9, 127.6, 127.1, 126.3, 125.6, 123.6, 121.7, 96.2, 88.9, 14.6. IR (ATR) 3024, 2919, 2853, 1593, 1486, 1444, 1414, 1378, 1251, 1152, 1069, 1025, 907, 809, 754, 682. MS (EI  $+)$  m/z = 426 (62), 411 (67), 349 (20), 334 (15), 289 (10), 206 (11), 149 (10), 136 (73), 111 (25), 97 (43), 69 (54), 57 (100). HR MS (EI +)  $m/z$  calcd for  $C_{31}H_{22}S$  ([M]<sup>+</sup>), 426.1442; found, 426.1440.

4-Methoxy-2′-((4-methoxyphenyl)ethynyl)-1,1′:3′,1″-terphenyl 9i. Yield: 62% (121 mg, 0.31 mmol), yellow oil. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.73–7.61 (m, 4H), 7.52–7.32 (m, 6H), 7.12–6.92 (m, 4H), 6.83−6.68 (m, 2H), 3.90 (s, 3H), 3.77 (s, 3H). 13C{1 H} NMR (75 MHz, CDCl<sub>3</sub>) δ 159.5, 159.2, 144.9, 144.3, 141.5, 133.8, 132.6, 131.1, 129.9, 128.5, 128.2, 127.8 (2C), 127.4, 120.7, 116.0, 114.0, 113.3, 96.0, 88.0, 55.5, 55.4. IR (ATR) 3053, 3002, 2957, 2834, 2211, 1604, 1571, 1507, 1450, 1288, 1243, 1174, 1027, 829, 803, 758, 698, 576, 529. MS  $(EI+)$   $m/z = 390$  (76,  $[M]^+$ ), 359 (51), 315 (27), 283 (56), 268 (26), 239 (100), 150 (56), 84 (8). HR MS (EI+)  $m/z$  calcd for  $C_{28}H_{22}O_2$ ([M]+ ), 390.1620; found, 390.1627.

2-Methoxy-2′-((2-methoxyphenyl)ethynyl)-1,1′:3′,1″-terphenyl **9j**. Yield: 72% (139 mg, 0.36 mmol), yellow oil. <sup>1</sup>H NMR (300 MHz, CDCl3) δ 7.81−7.72 (m, 2H), 7.52−7.31 (m, 8H), 7.14−7.00 (m, 3H), 6.87−6.64 (m, 3H), 3.78 (s, 3H), 3.60 (s, 3H). 13C{1 H} NMR (75 MHz, CDCl<sub>3</sub>) δ 160.0, 157.3, 144.1, 142.4, 141.3, 133.4, 131.7, 130.8, 130.0, 129.2, 129.1, 128.9, 128.5, 127.8, 127.7, 127.2, 122.2, 120.34, 120.32, 113.5, 111.2, 110.9, 92.9, 92.0, 55.9, 55.6. IR (ATR) 3059, 2926, 2841, 1595, 1494, 1460, 1432, 1270, 1245, 1270, 1245, 1157, 1103, 1027, 802, 753, 699. MS (EI+)  $m/z = 390$  (100, [M]<sup>+</sup>), 375 (28), 359 (51), 344(26), 326 (20), 313 (30), 283(24) 268 (13), 252 (13), 239 (21), 157 (14). HR MS (EI+)  $m/z$  calcd for  $C_{28}H_{22}O_2$  ([M]<sup>+</sup>), 390.1620; found, 390.1626.

Ethyl 2′-((4-(Ethoxycarbonyl)phenyl)ethynyl)-[1,1′:3′,1″-terphen $y$ ]]-4-carboxylate **9k**. Yield: 66% (157 mg, 0.33 mmol), yellow solid. mp 124−126 °C. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ 8.23−8.10 (m, 2H), 7.92−7.85 (m, 2H), 7.81−7.72 (m, 2H), 7.71−7.61 (m, 2H), 7.56− 7.36 (m, 6H), 7.03 (dd, J = 8.4, 1.7 Hz, 2H), 4.45 (q, J = 7.1 Hz, 2H), 4.35 (q,  $J = 7.1$  Hz, 2H), 1.45 (t,  $J = 7.1$  Hz, 3H), 1.37 (t,  $J = 7.1$  Hz, 3H).  ${}^{13}C{^1H}$  NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  166.7, 166.1, 145.8, 145.6, 144.2, 140.9, 131.0, 129.8, 129.7, 129.4, 129.3, 129.2, 128.9, 128.5, 128.0, 127.9, 127.8, 119.9, 95.6, 91.6, 61.2 (2C), 14.5, 14.4. IR (ATR) 3050, 2985, 2913, 1717, 1603, 1397, 1364, 1267, 1102, 1022, 852, 755, 693. HR MS (ESI+)  $m/z$  calcd for  $C_{32}H_{26}O_4$  ([M + H]<sup>+</sup>), 474.1831; found, 474.1842.

4-Bromo-4″-methoxy-2′-((4-methoxyphenyl)ethynyl)-1,1′:3′,1″ terphenyl 9l. Yield: 60% (141 mg, 0.30 mmol), brown solid. mp 112− 113 °C. <sup>1</sup> H NMR (300 MHz, CDCl3) δ 7.72−7.50 (m, 6H), 7.45−7.27 (m, 3H), 7.07−6.94 (m, 4H), 6.84−6.69 (m, 2H), 3.89 (s, 3H), 3.78 (s, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  159.6, 159.2, 144.4, 143.3, 140.3, 133.5, 132.5, 131.4, 130.9, 130.8, 128.9, 128.7, 127.8, 121.6, 120.4, 115.6, 113.9, 113.2, 96.3, 87.5, 55.4, 55.3. IR (ATR) 3049, 2998, 2924, 2832, 1604, 1505, 1449, 1386, 1290, 1241, 1172, 1103, 1070, 1033, 825, 791, 748, 523. MS (EI+) m/z = 468 (2, [M]<sup>+</sup> ), 433 (23), 419 (46), 404 (8), 389 (18), 374 (55), 357 (15), 341 (21), 327 (16), 281 (64), 207 (100), 73 (71). HR MS (EI+)  $m/z$  calcd for  $C_{28}H_{21}BrO_2$ ([M]+ ), 468.0725; found, 468.0730.

General Procedure for the Cobalt-Catalyzed Cross-Benzannulation of 1,3-Enynes with Unsymmetrical Buta-1,3-diynes. Under an argon atmosphere cobaltdibromo(1,3-bis- (diphenylphosphino)propane)  $[= \text{CoBr}_2(\text{dppp})]$  (10 mol %), zinc powder (20 mol %), and zinc iodide (20 mol %) were dissolved in  $CH_2Cl_2$  (2 mL/mmol). But-3-en-1-yn-1-ylbenzene (0.60 mmol, 1.20 equiv) and buta-1,3-diyn (0.50 mmol, 1.00 equiv) were added subsequently. The reaction was monitored by GC-MS analysis. After 50% conversion of the 1,3-enyne, additional but-3-en-1-yn-1-ylbenzene (0.3 equiv) was added. The reaction was stirred at room temperature

until complete conversion was determined by TLC and GC-MS analyses. After complete reaction, n-pentane was added, followed by filtration over a plug of silica (n-pentane/Et<sub>2</sub>O 1:1). The solvent was removed, and the crude mixture was purified by flash chromatography (silica, eluent: *n*-pentane/ $Et<sub>2</sub>O$ ).

2′-(Hept-1-yn-1-yl)-1,1′:3′,1″-terphenyl 10a and 3-Pentyl-2-(phenylethynyl)-1,1′-biphenyl 11a. Yield: 50% (81 mg, 0.25 mmol, 10a/  $11a = 61:39$ ), yellow oil. Analytical data of  $10a: {}^{1}H$  NMR (300 MHz, CDCl3) δ 7.67−7.55 (m, 2H), 7.47−7.29 (m, 4H), 7.29−7.10 (m, 7H), 2.93 (dd, J = 8.9, 6.5 Hz, 2H), 1.87−1.58 (m, 2H), 1.51−1.26 (m, 4H), 0.91 (t, J = 7.0 Hz, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  145.9, 144.8, 141.5, 131.3, 129.8, 128.4, 128.14, 128.08, 127.9, 127.8, 127.4, 127.1, 124.0, 121.2, 96.4, 88.2, 35.5, 32.1, 30.7, 22.8, 14.2. IR (ATR) 3057, 2925, 2857, 1943, 1878, 1806, 1597, 1567, 1492, 1455, 1377, 1249, 1177, 1108, 1070, 1026, 972, 913, 841, 801, 755, 694, 634, 623, 577, 520, 485, 411. MS (EI+) m/z = 234 (57, [M]<sup>+</sup> ), 267 (60), 256 (100), 252 (100), 239 (46), 233 (21), 215 (17), 203 (32), 178 (90), 165 (24), 57 (20), 32 (28). HR-MS (EI+)  $m/z$  calcd for  $C_{25}H_{24}$  $([M]^{\dagger})$ , 324.1878; found, 324.1865. Analytical data of 11a: <sup>1</sup>H NMR  $(300 \text{ MHz}, \text{CDCl}_3)$   $\delta$  7.69–7.54 (m, 4H), 7.47–7.28 (m, 9H), 2.08 (t, J = 6.9 Hz, 2H), 1.33−1.21 (m, 2H), 1.21−1.13 (m, 2H), 1.13−1.03 (m, 2H), 0.82 (t, J = 7.1 Hz, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$ 145.1, 141.6, 129.7, 128.5, 127.8, 127.32, 127.26, 121.2, 97.7, 79.3, 31.0, 27.9, 22.3, 19.6, 14.0. IR (ATR) 3057, 2927, 2860, 1599, 1569, 1495, 1454, 1422, 1072, 1028, 914, 807, 756, 697. MS (EI+) m/z = 234 (8, [M]<sup>+</sup>), 281 (15), 267 (30), 253 (100), 239 (13), 207 (25). HR MS (EI +)  $m/z$  calcd for  $C_{25}H_{24}$  ([M]<sup>+</sup>), 324.1878; found, 324.1881.

([1,1′:3′,1″-Terphenyl]-2′-ylethynyl)trimethylsilane 10b and Trimethyl(2-(phenylethynyl)-[1,1′-biphenyl]-3-yl)silane 11b. Yield: 46% (75 mg, 0.23 mmol,  $10b/11b = 82:18$ ), yellow oil. <sup>1</sup>H NMR  $(300 \text{ MHz}, \text{CDCl}_3)$   $\delta$  7.65−7.60 (m, 3H), 7.46−7.32 (m, 10H), −0.04  $(s, J = 3.1 \text{ Hz}, 9\text{H})$ . Analytical data of the resolved signals of the major isomer 10b: <sup>13</sup>C{<sup>1</sup>H} NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  145.6, 141.1, 129.8, 128.5, 128.3, 127.8, 127.5, 120.4, 103.8, 102.1, −0.5. Analytical data of the resolved signals of the minor isomer  $\mathbf{11b:} \mathrm{^{13}C(^{1}H)}$  NMR (75 MHz, CDCl3) δ 130.1, 129.9, 128.6, 128.1, 120.4, −0.5. IR (ATR) 3057, 2154, 1600, 1569, 1449, 1412, 1248, 1209, 1074, 1027, 915, 839, 753, 695, 641, 607, 556, 516, 448. MS (EI+)  $m/z = 326$  (43, [M]<sup>+</sup>), 311 (26), 295 (68), 279 (8), 267 (31), 252 (100), 226 (5), 155 (4), 73 (11). HR MS (EI+)  $m/z$  calcd for  $C_{23}H_{22}Si$  ([M]<sup>+</sup>), 326.1491; found, 326.1490.

Trimethyl((3-pentyl-[1,1′-biphenyl]-2-yl)ethynyl)silane 10c and Trimethyl((3-pentyl-[1,1′-biphenyl]-2-yl)ethynyl)silane 11c. Yield: 56% (90 mg, 0.28 mmol, 10c/11c = 80:20). The major product could be separated after additional column chromatography (eluent: npentane) as a colorless oil (71 mg, 0.22 mmol, 44%). <sup>I</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.65–7.55 (m, 2H), 7.46–7.31 (m, 3H), 7.31–7.25 (m, 1H), 7.21 (ddd, J = 6.7, 5.2, 1.7 Hz, 2H), 2.96−2.76 (m, 2H), 1.82−1.61 (m, 2H), 1.49−1.31 (m, 4H), 0.95 (t, J = 7.3 Hz, 3H), 0.15  $(s, 9H)$ . <sup>13</sup>C{<sup>1</sup>H} NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  146.5, 145.0, 141.2, 129.7, 128.2, 127.71, 127.66, 127.3, 127.0, 121.2, 103.2, 101.9, 35.5, 32.1, 30.6, 22.7, 14.2, −0.2. IR (ATR) 2956, 2927, 2860, 2152, 1457, 1430, 1249, 841, 755, 698, 639, 608. MS (EI+)  $m/z = 320$  (57, [M]<sup>+</sup>), 305 (41), 246 (100), 231 (64), 217 (89), 212 (67), 191 (84), 165 (11), 73 (52). HR MS (EI+)  $m/z$  calcd for  $C_{22}H_{28}Si$  ([M]<sup>+</sup>), 320.1960; found, 320.1968.

((2-Methoxy-[1,1′:3′,1″-terphenyl]-2′-yl)ethynyl)trimethyl-silane 10d. Yield: 69% (123 mg, 0.345 mmol), yellow oil. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.72–7.63 (m, 2H), 7.47–7.30 (m, 8H), 7.09–6.94 (m, 2H), 3.81 (s, 3H), −0.06 (s, 9H). 13C{1 H} NMR (75 MHz, CDCl3) δ 157.1, 144.6, 143.1, 141.0, 131.4, 130.6, 129.8, 129.0, 128.9, 128.3, 128.0, 127.7, 127.3, 121.8, 120.2, 111.0, 103.9, 100.7, 55.7, −0.3. IR (ATR) 3059, 2957, 2835, 2156, 1947, 1597, 1496, 1459, 1418, 1246, 1179, 1106, 1030, 863, 755, 699, 642, 612, 567, 488, 425. MS (EI+) m/  $z = 356 (100, [M]^+), 355 (82), 341 (45), 309 (55), 293 (27), 283 (49),$ 265 (66), 252 (52), 239 (34), 89 (37), 73 (44), 59 (35). HR MS (EI+) *m/z* calcd for  $C_{24}H_{24}OSi$  ([M]<sup>+</sup>), 356.1596; found, 356.1589.

((4-Methoxy-[1,1′:3′,1″-terphenyl]-2′-yl)ethynyl)trimethyl-silane 10e and (2-((4-Methoxyphenyl)ethynyl)-[1,1′-biphenyl]-3-yl) trimethylsilane 11e. Yield:  $61\%$  (109 mg, 0.305 mmol,  $10e/11e =$ 

<span id="page-5-0"></span>91:9). Analytical data of major isomer 10e: <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.67–7.57 (m, 4H), 7.49–7.30 (m, 6H), 7.04–6.88 (m, 2H), 3.88 (s, 3H), 0.02 (s, 9H). <sup>13</sup>C{<sup>1</sup>H} NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  159.2, 145.7, 145.1, 141.1, 133.5, 130.9, 129.7, 128.4, 128.3, 128.1, 127.7, 127.4, 120.2, 113.2, 104.0, 101.9, 55.4, −0.43. IR (ATR) 3288, 3056, 2835, 2153, 1946, 1884, 1608, 1575, 1512, 1453, 1290, 1246, 1178,1110, 1076, 1034, 839, 805, 759, 699, 638, 604, 571, 536, 442. MS (EI+) m/z = 356 (100, [M]<sup>+</sup> ), 325 (29), 283 (62), 268 (21), 252 (46), 239 (76), 73 (15), 59 (15), 32 (100). HR MS (EI+) m/z calcd for  $C_{24}H_{24}OSi$  ([M]<sup>+</sup>), 356.1596; found, 356.1599.

((4-Bromo-[1,1′:3′,1″-terphenyl]-2′-yl)ethynyl)trimethyl-silane 10f and (2-((4-Bromophenyl)ethynyl)-[1,1′-biphenyl]-3-yl) trimethylsilane 11f. Yield: 60% (122 mg, 0.30 mmol, 10f/11f = 85:15). <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ 7.66–7.59 (m, 2H), 7.58–7.50 (m, 3H), 7.47−7.36 (m, 6H), 7.33 (dd, J = 7.2, 1.8 Hz, 1H), −0.09 (s, 9H). Resolved signals of major isomer  ${\bf 10f:}~^{13}\rm C\{^1H\}$  NMR (75 MHz, CDCl3) δ 145.6, 144.2, 140.9, 140.0, 131.5, 130.9, 129.7, 128.8, 128.5, 128.2, 127.8, 127.6, 121.8, 120.3, 103.4, 102.7, −0.5. Resolved signals of minor isomer 11f:  ${}^{13}C(^{1}H)$  NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  130.8, 129.8, 128.3, 127.8, 127.5, 120.2, 100.1. IR (ATR) 2958, 2153, 1491, 1451, 1418, 1388, 1248, 1072, 1009, 835, 804, 754, 697, 650, 602, 515, 449, 415. MS (EI+)  $m/z = 404$  (30, [M]<sup>+</sup>), 391 (21), 373 (9), 332 (10), 309 (12), 295 (100), 279 (9), 265 (23), 252 (73), 73 (16). HR MS (EI+)  $m/z$  calcd for  $C_{23}H_{21}BrSi$  ([M]<sup>+</sup>), 404.0596; found, 404.0596.

## ■ ASSOCIATED CONTENT

## **6** Supporting Information

Detailed X-ray structure of compound  $9h$  and copies of  $H$  and <sup>13</sup>C NMR spectra of 4a–f, 7a–g, 7l, 8a–d, 9a–l, and 10a/11a– 10f/11f. The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/ acs.joc.5b01198.

#### ■ [AUTH](http://pubs.acs.org/doi/abs/10.1021/acs.joc.5b01198)[OR](http://pubs.acs.org)[INFORMATION](http://pubs.acs.org)

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#### Notes

The aut[hors declare no competing](mailto:hilt@chemie.uni-marburg.de) financial interest.

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